HPLC Separation of Glucosinolates from Leaves and Seeds of *Arabidopsis thaliana* and Their Identification Using Thermospray Liquid Chromatography/Mass Spectrometry*

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Abstract

Leaf and seed extracts of Arabidopsis thaliana var. Columbia contain a large number of glucosinolates, representing close to 25% of those known to occur in nature. The glucosinolates, in the form of their desulphated analogs, are separated by reversed-phase, high-performance liquid chromatography (HPLC). Seventeen are identified using thermospray liquid chromatography/mass spectrometry (TSP LC/MS). Additional glucosinolates, present in trace amounts, are identified as isothiocyanates by electron impact and chemical ionization gas chromatography/MS (GC/MS). In total, 23 glucosinolates are detected and these include four series of homologs and analogs. Fifteen possess aliphatic side chains, of which six contain ω methylthioalkyl and six contain ω -methylsulphinylalkyl side chains; eight possess aromatic side chains, of which four constitute an homologous series of benzoic acid esters and three possess 3-indolylmethyl-based structures. Sixteen of the glucosinolates are detected in Arabidopsis thaliana for the first time and three of these, 4-hydroxybutyl glucosinolate, 5-benzoyloxypentyl glucosinolate, and 6-benzoyloxyhexyl glucosinolate, represent novel plant constituents.

Introduction

More than 100 glucosinolates are known. They are represented by the general structure:



In radish, esterification of the thioglucosyl moiety by sinapic acid is also known to occur (1,2). All can be hydrolyzed by the enzyme myrosinase to give D-glucose, sulphate, and a variety of products derived from the aglycone moiety, including isothiocyanates, nitriles, and other products (3). Glucosinolates have been reported in 11 plant families and within the Cruciferae they have been found in all species examined (4).

Glucosinolates, as a class of plant constituents, are of major importance to several vegetable crops due to the flavor components (isothiocyanates or mustard oils) generated by hydrolysis when the plant tissues are crushed. The hydrolytic products of some glucosinolates exhibit toxic effects in mammals and, thus, have limited the usefulness of some Cruciferae species as fodder (1). Information concerning the biosynthetic pathway of glucosinolates and the genes that encode the enzymes involved in their synthesis could facilitate the genetic engineering of altered glucosinolate levels in crops. One approach to obtaining such information is through the analysis of mutants. As part of a program on the genetic manipulation of glucosinolate levels in crop plants, the present authors plan to take advantage of the speed and versatility of genetic analysis in the small crucifer Arabidopsis thaliana (L.) Heynh to isolate and characterize mutants deficient in glucosinolate biosynthesis (5). A necessary first step was to determine the natural composition of glucosinolates in leaf and seed tissue of wild type plants and to adopt a method that would be amenable for their analysis in individual plants.

Only fragmentary reports are available on the glucosinolate composition of seed and leaf tissue of *Arabidopsis*. In one study, allylglucosinolate and isopropylglucosinolate were identified in leaf tissue as enzyme-derived isothiocyanates and nitriles (6). Seven different glucosinolates have been detected in extracts of seeds by employing negative ion fast atom bombardment (FAB) mass spectrometry (7) and chemical ionization/mass spectrometry (CI/MS) (8,9). In a preliminary highperformance liquid chromatographic (HPLC) analysis of seed and leaf extracts of *Arabidopsis*, the present authors found evidence of an even greater number of glucosinolates, and only a few exhibited retention times that corresponded with available reference standards.

The separation of glucosinolates present in wild type Arabidopsis thaliana leaves and seeds by reversed-phase HPLC of their desulphated analogs is reported here. The technique is rapid and allows detection of the predominant glucosinolates using quantities of leaf and seed tissue that can be obtained from a single plant. Glucosinolates were identified using the methods of thermospray liquid chromatography/MS (TSP LC/MS) (10,11) and gas chromatography (GC)/MS of their myrosinasederived isothiocyanates.

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Experimental

Arabidopsis thaliana (L.) Heynh. used in this study were descended from the Columbia wild type. The methods employed for growing the plants and harvesting seed were as described by Somerville and Ogren (12), except an 18-h photoperiod rather than continuous illumination was employed.

Glucosinolates were recovered from leaves and seeds separately by extraction into 20 times their fresh weight of boiling 80% ethanol. In the case of leaves, the tissue was rapidly weighed following removal from the plant and immediately placed into boiling solvent. Seeds and leaves were extracted for ca. 5 min at this temperature and ground using a Polytron tissue homogenizer; then, the mixtures were filtered. The residues were extracted again in a similar volume of boiling 80% ethanol. Ethanol was removed from each combined extract by rotary evaporation at 35°C, and the aqueous mixture was filtered through a Celite pad to give a clear solution.

An aliquot of the glucosinolate solution was adjusted to pH 7 with phosphate buffer and extracted three times, each time with two volumes of CH_2Cl_2 . After removing traces of solvent, the aqueous solution was incubated with myrosinase for 2 h at 20°C, and the liberated isothiocyanates were recovered by three extractions, each time with two volumes of CH_2Cl_2 . The solvent was dried (MgSO₄) and concentrated by rotary evaporation at 35°C.

Another portion of the glucosinolate solution was applied to DEAE-Sephadex A-25 (pyridine acetate form), and the glucosinolates were converted into their desulpho analogs by overnight treatment with aryl sulphatase (Sigma Chemical, Type H-1). The procedure used was similar to that described by Mitchinton et al. (13). The column eluate containing the desulphated products was placed in a boiling water bath for 3 min to inactivate any β -galactosidase activity, and then it was concentrated in a Savant Speed Vac concentrator. Ortho-nitrophenyl- β -D-galactopyranoside was added as an internal standard when required, and the samples were passed through a 0.22- μ m filter.

HPLC of desulfoglucosinolates was carried out using a Spectra Physics Model 8700 XR system equipped with a Model 8780 XR autosampler and a Spectroflow 773 UV detector at a wavelength setting of 226 nm, 0.1 AUFS. Chromatograms were recorded using a Hewlett-Packard Model 3392A plotter-integrator. Samples, 50 μ L each, were separated using a Whatman Partisphere C₁₈ column (110×4.7 mm i.d., 5- μ m particle size). Mobile phase at 1 mL/min was programmed as follows: 1.25% acetonitrile in water for 5 min, a linear gradient from 1.25 to 22.5% acetonitrile over the next 15 min, constant at 22.5% acetonitrile for 5 min, and a linear gradient to 100% acetonitrile for 15 min. The column was reconditioned by a linear gradient to 1.25% acetonitrile over 5 min and washed with this concentration for 15 min prior to an injection.

LC separation of desulphoglucosinolates in TSP LC/MS was performed using a Waters system consisting of two pumps, Models 590 and 510, controlled by a Model 680 gradient controller. Samples, 20 μ L each, were injected, and separation of the desulfoglucosinolates was achieved on a C₁₈ column similar to the one indicated above, but using a linear gradient elution with 0.1 M ammonium acetate and acetonitrile (0 to 60% acetonitrile over 30 min) at a flow rate of 0.9 mL/min. TSP LC/MS was performed as described (11) employing a Model 3300 Finnigan mass spectrometer retrofitted with a Vestec thermospray interface. This mass spectrometer and the Model 4500 used for GC were interfaced to an Incos 2300 data acquisition system (Finnigan). The thermospray source was maintained at 265°C, and spectra were acquired by scanning from m/z 115 to 450 every 2.0 s.

GC/MS data for isothiocyanates were obtained using a Finnigan Model 4500 GC-MS. CI (methane reagent and helium carrier gas) and electron impact (EI) (70 eV ionization energy) mass spectra were acquired every second over the mass range of 40 to 400. Samples were separated using a DB-5 (J&W Scientific) phenyl methyl silicone fused-silica column (60 m \times 0.32 mm i.d.). After on-column injection with the injector cooled to 38°C, the oven temperature was raised with maximum heating from 40° to 100°C and then programmed at 4°/min to 310°C.

Results and Discussion

Although glucosinolates themselves constitute a uniform class of natural compounds, they do not all yield similar myrosinase hydrolysis products that can be readily analyzed; therefore, the method of reversed-phase HPLC (13) was utilized for their analysis in Arabidopsis seed and leaf tissues. The HPLC profiles of desulfoglucosinolates obtained (Figure 1) indicated the presence of a large number of components. Their identities are listed in Table I and discussed below; each is designated numerically, and these numbers are employed in all figures and tables. Both the composition and relative amounts of the glucosinolates in the two tissue extracts differed appreciably. That the major peaks in the UV trace of the HPLC effluents were, in fact, desulfoglucosinolates was confirmed in a parallel experiment. Extracts were treated in the same way as those initially employed, except myrosinase was added to the crude glucosinolate solution prior to its treatment with sulfatase and recovery on the Sephadex column. Apart from the presence of one or two peaks eluting within the first 1.6 min of the chromatogram, none of the major peaks recorded in the initial chromatograms were found in the myrosinase-treated seed and leaf extracts.

Recent successes in qualitative analyses of glucosinolates as their desulphated analogs using TSP LC/MS (10,11) led to the use of this method for the identification of the glucosinolates detected in the leaf and seed extracts. Positive ion TSP spectra





	Structure of		Structure of
Aliphatic glucosinolates	K group	Aromatic giucosinolates	к group
With Methylthioalkyl side chains 1. 3-Methylthiopropyl 2. 4-Methylthiobutyl 3. 5-Methylthiopentyl 4. 6-Methylthiobexyl	CH ₃ -S-(CH ₂) ₃ - CH ₃ -S-(CH ₂) ₄ - CH ₃ -S-(CH ₂) ₅ - CH ₃ -S-(CH ₂) ₅ -	With heterocyclic side chains 16. 3-Indolylmethyl	COLUCE CH2-
6. 8-Methylthiooctyl	CH ₃ -S-(CH ₂) ₇ - CH ₃ -S-(CH ₂) ₈ -	17. 4-Methoxy-3-indolylmethyl	осн _а
With Methylsulphinylalkyl side chains 7. 3-Methylsulphinylpropyl 8. 4-Methylsulphinylbutyl 9. 5-Methylsulphinylpentyl 10. 6-Methylsulphinylhexyl 11. 7-Methylsulphinylheptyl 12. 8-Methylsulphinyloctyl	CH ₃ -SO-(CH ₂) ₃ - CH ₃ -SO-(CH ₂) ₄ - CH ₃ -SO-(CH ₂) ₅ - CH ₃ -SO-(CH ₂) ₅ - CH ₃ -SO-(CH ₂) ₇ - CH ₃ -SO-(CH ₂) ₈ -	18. 1-Methoxy-3-indolylmethyl	
With other side chains 13. 3-Hydroxypropyl 14. 4-Hydroxybutyl 15. 3-Butenyl	HO-(CH ₂) ₃ - HO-(CH ₂) ₄ - CH ₂ -CH-(CH ₂) ₂ -	With non-heterocyclic side chains 19. 3-Benzoyloxypropyl 20. 4-Benzoyloxybutyl 21. 5-Benzoyloxypentyl 22. 6-Benzoyloxyhexyl 23. 2-Phenylethyl	Ссн ₃ С ₆ H ₅ COO-(CH ₂) ₃ - С ₆ H ₅ COO-(CH ₂) ₄ - С ₆ H ₅ COO-(CH ₂) ₅ - С ₆ H ₅ COO-(CH ₂) ₆ - С ₆ H ₅ -(CH ₂) ₂ -

of desulphoglucosinolates contained prominent diagnostic ions from which the molecular structures of the glucosinolate side chains, R, can be deduced. It should be noted that structural assignments for these diagnostic ions have not been confirmed experimentally; however, the following structures are consistent with ions which have been observed (10,11):

$$[R-C=NOH]^+$$
 $[R-C=NOH+H]^+$ $[R-C=NOH+2H]^+$
(A) (A+1) (A+2)

$$[R - N = C = O + NH_4]^+ [R - C(-S) = NOH + 2H]^+ [M + H]^+$$

(B) (C) (D)

Ions of high intensities, often constituting the base peaks and corresponding to $[R]^+$, are found in the spectra of 3-indolylmethyldesulphoglucosinolate and its hydroxylated and methoxylated analogs (11, and unpublished data), making these compounds particularly suited for analysis using TSP LC/MS. Additionally, TSP spectra of desulphoglucosinolates are characterized by the presence of ions at m/z 180 and 214, which are related to the thioglucosyl portion of the molecules (10,11).

Seventeen glucosinolates were identified as their desulphated analogs by positive ion TSP LC/MS. The reconstructed ion current (RIC) chromatograms of leaf and seed extracts are presented in Figure 2. The differences in the elution times of the components in the two analyses were due to the employment of two different C_{18} columns. Further, the apparent low abundance of **6** in the seed TSP LC/MS trace (compare **6** in Figures 1 and 2) was an anomaly caused by an increase in the thermospray probe tip temperature as the organic content of the mobile phase increased during the gradient elution, which, in turn, decreased detection sensitivity.

Present in each desulphoglucosinolate spectrum were ions at m/z 180 and 214, indicative of the thioglucosyl moiety (Figures 3 through 5). Three indole-containing glucosinolates were readily identified by the complete correspondence of their spectra and retention times with those previously reported (11); these included 3-indolylmethylglucosinolate (16) and its 4-methoxy- and



1-methoxy-substituted analogs (17,18). An homologous series of five ω -methylthioalkylglucosinolates (2 to 6) was identified by diagnostic fragment ions and the appearance of protonated molecular ions, D, at m/z 342, 356, 370, 384, and 400; the spectra of these components are given in Figure 3. Ions corresponding to A, A + 1, A + 2, B, B + 1, and C were present in the spectrum of each component. Further, the homologous nature of this series was indicated by their elution times (Figures 1 and 2) The identification of three other desulphoglucosinolates (14, 19, and 20) also was based on the appearance of a similar series of diagnostic fragmentation and molecular ions (Figure 4). The spectrum of 4-hydroxybutyl desulphoglucosinolate (14) was dominated by the ion at m/z 137 corresponding to structure A + 1, and although no ion was observed for A, ions representing B, B + 1, and C were present.

The remaining group of components, whose structures were initially deduced from their TSP spectra, constituted another

homologous series, ω -methylsulphinylalkylglucosinolates (7 to 12). Although members of this group were the more abundant components in leaf and seed extracts (Figure 2), protonated molecular ions were not detected (Figure 5). The spectrum of 3-methylsulphinylpropyl desulphoglucosinolate (7, n = 3) contains ions corresponding to each of the three fragmentation groups (A through C). In addition, the spectrum contains two ions, both of low intensity at 26 and 16 mass units, less than that assigned fragmentation ion A. As the length of the side chain increased from n = 3 to n = 8, the ion intensity of A + 1 decreased; A increased to the point where it represented the

base peak in the spectra of 10, 11, and 12; fragmentation ions B and C decreased to the point where C was no longer observed in spectra where n > 5.

Structures of glucosinolates identified by TSP LC/MS, except for the 3 indole-containing group, were reconfirmed by El or CI GC/MS of isothiocyanates formed by treating the glucosides with myrosinase. (Note the same numbers that have been assigned to designate glucosinolate structures apply also to their derived isothiocyanates.) Six other glucosinolates not in sufficient concentration in either the leaf or seed extracts to be detectable by TSP LC/MS were identified. The six were 3-methylthiopropyl (1), 3-hydroxypropyl (13), 3-butenyl (15),







Figure 5. Thermospray LC/MS spectra of ω -methylsulphinylalkyl desulphoglucosinolates. Underlined numbers refer to glucosinolates listed in Table I.



5-benzoyloxypentyl (21), 6-benzoyloxyhexyl (22), and 2-phenylethyl (23) glucosinolates. The RICs (EI/MS) of the leaf- and seed-derived isothiocyanates are shown in Figure 6. The spectra of the ω -methylthioalkyl series (1 to 6), the ω methylsulphinylalkyl series (7 to 9), 3-hydroxypropyl isothiocyanate (13), 3-butenyl isothiocyanate (15), 3-benzoyloxypropyl isothiocyanate (19), and 2-phenylethyl isothiocyanate (23) corresponded exactly with spectra previously reported for these compounds (14–16). In addition, EI/MS data (m/z (% relative abundance)) were obtained as follows and are consistent with the indicated compounds:

4-hydroxybutyl isothiocyanate (14)

131 (M^{+•}, 59), 113 (9.7), 98 (18), 85 (82.4), 75 (14.3),

74 (10.2), 72 (75.9), 71 (43.6), 70 (39.8), 62 (46),

60 (29.2), and 55 (100)

4-benzoyloxybutyl isothiocyanate (20) 235 (M^{+•}, 5.8) 113 (29.8), 105 (100), 85 (64.4), 77 (49), 72 (15.9), 55 (20.8), and 51 (19.9)

5-benzoyloxypentyl isothiocyanate (21)

249 (M^{+•}, 1.8), 127 (43.2), 105 (100), 85 (20.6), 77

(55.9), 72 (15.8), 69 (29.5), 68 (68.8), and 51 (21.9)

6-benzoyloxyhexyl isothiocyanate (22)

263 (M^{+•}, 4.8), 141 (17.9), 105 (100), 85 (22.4), 83

(26.9), 77 (49.1), 72 (17.2), 55 (30.1) and 51 (19.4)

Although EI spectra were obtained for the ω -methylsulphinyl isothiocyanates derived from 7 to 9, the spectra of higher members of the series were not similarly attainable. This was due to the extensive thermal degradation of this group during GC/MS and evidenced in the base line undulations in the traces given in Figure 6. Similar degradation occurred during CI GC/MS; however, spectra consistent with the complete series 7 to 12 were attained. Ions at m/z (relative abundance) 204 (1.5) $[M+41]^+$, 192 (6.8) $[M+29]^+$, 164 (71) $[M+1]^+$, 148 (7.9) $[(M+1)-16]^+$, 105 (100), 100 (8.5), and 89 (20) constituted the spectrum of 3-methylsulphinylpropyl isothiocyanate. A similar series of ions, each increased by 14 mass units, was found for each higher homolog of the series 8 to 12. As the chain lengths increased, so also did the relative abundance of the ions in the range $[(M+1)-16]^+$ to $[M+41]^+$. Further confirmatory data of the structure of 12 was provided when this component was isolated as its desulpho analog by collecting the appropriate HPLC effluent fraction of a seed extract. Analyzed by FAB/MS (VG model 70-250 SEQ mass spectrometer), an exact mass of 414.1637 (C16H32O7NS2) was obtained, which is consistent with the protonated molecular ion of 8-methylsulphinyloctyl desulphoglucosinolate.

Athough the glucosinolates were not quantitated in this study, it is evident that major differences exist in the relative proportions of the individual glucosinolates found in leaf and seed tissue (Figures 1 and 2). This is exemplified by the two methoxy analogs (17, 18) of 3-indolylmethyl glucosinolate that were found in leaf tissue, but not in seeds. Sang et al. (17) also have observed a similar tissue distribution of indole-containing glucosinolates in a number of Brassica spp. Leaf tissue extracts contain relatively large concentrations of the ω -methylsulphinylalkyl glucosinolates 7, 8, 9, and 12. In contrast, seed tissues possess comparatively large amounts of the ω -methylthioalkyl series 2 to 6 and the aromatic esters 19 and 20. Nevertheless, most of the 23 glucosinolates identified were found in both tissue extracts, although some could only be detected in trace amounts by GC/MS of their isothiocyanates using concentrated solutions. Concerning the two ω -hydroxyalkyl glucosinolates 13 and 14, although it is possible that they may have been derived by hydrolysis from 19 and 20, respectively, considerable care was taken during plant workup to avoid this occurrence (16).

Conclusion

TSP LC/MS has been found to be an excellent method for the separation and identification of the glucosinolates in *Arabidopsis thaliana*. Glucosinolates such as those containing sulfoxide or indolyl functions, which are not easily analyzed using GC/MS techniques, were identified without difficulty. The structures of all of the TSP-identified glucosinolates that yield isothiocyanates as products of myrosinase hydrolysis were confirmed by GC/MS. GC/MS analysis of isothiocyanates, although not applicable to all glucosinolates, was more sensitive than TSP LC/MS, allowing for the detection of trace amounts of glucosinolates. Of the 23 glucosinolates detected in this study, 16 constitute newly identified components of *Arabidopsis thaliana*, and of these, three glucosinolates, **14**, **21**, and **22**, have not been reported to occur in plants. The presence of seven glucosinolates previously indicated (7–9) as constituents of seeds were confirmed; these were glucosinolates **2**, **5**, **6**, **11**, **12**, **19**, and **20**. The presence of three, allylglucosinolate, isopropylglucosinolate (6), and 6-heptenylglucosinolate (7), whose occurrence had been noted earlier, were unverified. Indeed, the lack of detection of these compounds could be due to varietal differences within this species.

The number and diversity in structure of the glucosinolates found in *Arabidopsis thaliana* provide an unusual opportunity for studies of the biochemical genetics of this class of sulphurcontaining glucosides.

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